organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Graham Smith,* Steven E. Bottle, Damien A. Reid and Raymond C. Bott

Centre for Instrumental and Developmental Chemistry, Queensland University of Technology, GPO Box 2434, Brisbane 4001, Australia

Correspondence e-mail: g.smith@qut.edu.au

Key indicators

Single-crystal X-ray study T = 295 K Mean σ (C–C) = 0.007 Å R factor = 0.039 wR factor = 0.125 Data-to-parameter ratio = 7.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

An isoindoline EPR label: 5-acetamido-1,1,3,3-tetramethylisoindolin-2-yloxyl

The crystal structure of the isoindoline nitroxide amide 5-acetamido-1,1,3,3-tetramethylisoindolin-2-yloxyl, $C_{14}H_{19}$ - N_2O_2 , which has utility as an EPR label, shows a characteristically stable tetramethyl-substituted nitroxide ring system which is essentially coplanar with the amide side-chain substituent. A single intermolecular hydrogen bond between the amide H atom and the nitroxide O atom links the molecules head-to-tail into an infinite polymeric chain.

Received 6 June 2001 Accepted 27 June 2001 Online 27 July 2001

Comment

Nitroxides are often utilized as reporter molecules and, as such, are commonly referred to as spin labels or spin probes depending upon whether they are covalently linked (labels), or not (probes), to the system being studied. Precise knowledge of the bonding structure of spin labels is essential to obtain information on complex systems such as enzymes and other macromolecules (Farrens et al., 1996; Steinhoff et al., 1997; Rink et al., 1997; Mchaourab et al., 1997). Another crucial issue in the utilization of stable nitroxide probes in biological systems is the partitioning behaviour which governs membrane transfer. Charged nitroxides generally do not cross cell membranes, although there are exceptions (Kocherginsky & Swartz, 1995) and amphiphilic nitroxides may orient themselves with the non-polar portion of the molecule embedded in the lipid bilayer. Lipophilic nitroxides have also been associated with enhanced cytotoxicity, albeit in high concentrations (> 1 mM). We have synthesized (Reid *et al.*, 1998) amino substituted isoindoline nitroxides as potential new EPR spin labels/probes and in order to contrast the toxicity and partitioning preferences, have now synthesized the novel 5-acetamido-1,1,3,3-tetramethylisoindolin-2-yloxyl, (I), reported here.



Single-crystal X-ray analysis of the structure of this amide (Fig. 1) provides insight into the bonding characteristics resulting when this type of EPR label is bound to an amino acid sequence. This reveals a basic tetramethylisoindoline core which is relatively inflexible and therefore similar to the

 \odot 2001 International Union of Crystallography Printed in Great Britain – all rights reserved

previously reported compounds of the same type (Micallef et al., 1999). The N8–O8 bond distance [1.279 (5) Å] compares closely with these examples. The amide side chain adopts a conformation such that it is almost coplanar with the parent aromatic ring system [torsion angles C4-C3-N31-C31 $172.9 (5)^{\circ}$ and C3-N31-C31-O31 -5.5 (9)°]. This is stabilized by the presence of an intramolecular hydrogen bond between the amide-O and a ring-H atom [O31···H2-C2 2.880 (7) Å]. A single head-to-tail hydrogen bond between the amide group and the nitroxide O atom [N31-H31···O8¹ 2.884 (7) Å and N-H···O, 171 (3)° [symmetry code: (i) $-\frac{1}{2} + x$, $\frac{3}{2} - y$, $-\frac{1}{2} + z$] links the molecules into an infinite poymeric chain.

Experimental

The title compound was prepared according to the method of Reid et al. (1998), by reacting 5-amino-1,1,3,3-tetramethylisoindolin-2-yloxyl in tetrahydrafuran containing 1.5 equivalents of sodium bicarbonate, with 10 equivalents of acetyl chloride. Extraction into diethyl ether and recrystallization from acetonitrile gave data quality crystals.

Crystal data

$C_{14}H_{19}N_2O_2$ $M_r = 247.32$ Monoclinic, Cc a = 14.325 (4) Å b = 8.536 (8) Å c = 11.513 (4) Å $\beta = 91.56$ (2)° V = 1407 (1) Å ³ Z = 4	$D_x = 1.167 \text{ Mg m}^{-3}$ Mo K\$\alpha\$ radiation Cell parameters from 25 reflections $\theta = 10.4-16.3^{\circ}$ $\mu = 0.08 \text{ mm}^{-1}$ T = 295 K Block, yellow $0.38 \times 0.28 \times 0.15 \text{ mm}$
Data collection	
Rigaku AFC-7 <i>R</i> diffractometer ω -2 θ scans 1384 measured reflections 1286 independent reflections 716 reflections with $I > 2\sigma(I)$ $R_{int} = 0.020$ $\theta_{max} = 25.0^{\circ}$	$h = 0 \rightarrow 16$ $k = 0 \rightarrow 10$ $l = -13 \rightarrow 13$ 3 standard reflections every 150 reflections intensity decay: 2.6%
Refinement	
Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.039$ $wR(F^2) = 0.125$ S = 1.03 1286 reflections 164 parameters H-atom parameters constrained	$\begin{split} w &= 1/[\sigma^2(F_o^{-2}) + (0.0598P)^2 \\ &+ 0.1927P] \\ \text{where } P &= (F_o^{-2} + 2F_c^{-2})/3 \\ (\Delta/\sigma)_{\text{max}} &= 0.007 \\ \Delta\rho_{\text{max}} &= 0.20 \text{ e } \text{\AA}^{-3} \\ \Delta\rho_{\text{min}} &= -0.16 \text{ e } \text{\AA}^{-3} \\ \text{Extinction correction: } SHELXL97 \\ (\text{Sheldrick, 1997}) \\ \text{Extinction coefficient: } 0.0023 (2) \end{split}$

Data collection: MSC/AFC Diffractometer Control Software (Molecular Structure Corporation, 1999a); cell refinement: MSC/ AFC Diffractometer Control Software; data reduction: TEXSAN for



Figure 1

The molecular conformation and atom-naming scheme with the atoms shown as 30% probability ellipsoids (Spek, 1999).

Windows (Molecular Structure Corporation, 1999b); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); software used to prepare material for publication: TEXSAN for Windows (Molecular Structure Corporation, 1999b) and PLATON for Windows (Spek, 1999).

The authors acknowledge financial support from The Centre for Instrumental and Developmental Chemistry (Queensland University of Technology), and the Australian Research Council. Dr Peter Healy is thanked for collection of the X-ray diffraction data.

References

- Farrens, D. L., Altenbach, C., Yang, K., Hubbell, W. L. & Khorana, H. G. (1996). Science, 274, 768-770.
- Kocherginsky, N. & Swartz, H. M. (1995). In Nitroxide Spin Labels: Reactions in Biology and Chemistry, edited by N. Kocherginsky & H. M. Swartz, pp. 21-22. Baton Rouge: CRC Press.
- Mchaourab, H. S., Oh, K. J., Fang, C. J. & Hubbell, W. L. (1997). Biochemistry, 36 307-316
- Micallef, A. S., Bott, R. C., Bottle, S. E., Smith, G., White, J. M., Matsuda, K. & Iwamura, H. (1999). J. Chem. Soc. Perkin Trans. 2, pp. 65-71.

Molecular Structure Corporation (1999a). MSC/AFC Diffractometer Control Software. MSC, 9009 New Trails Drive, The Woodlands, TX 77381, USA.

Molecular Structure Corporation (1999b). TEXSAN for Windows. Version 1.06. MSC, 9009 New Trails Drive, The Woodlands, TX 77381, USA.

- Reid, D. A., Bottle, S. E. & Micallef, A. S. (1998). Chem. Commun. pp. 1907-1908.
- Rink, T., Riesle, J., Oesterhelt, D., Gerwert, K. & Steinhoff, H. J. (1997). Biophys. J. 73, 983-993.
- Sheldrick, G. M. (1997). SHELXL97 and SHELXS97. University of Göttingen, Germany.
- Spek, A. L. (1999). PLATON for Windows. September 1999 version. University of Utrecht, The Netherlands.
- Steinhoff, H. J., Radzwill, N., Thevis, W., Lenz, V., Brandenburg, D., Antson, A., Dodson, G. & Wollmer, A. (1997). Biophys. J. 73, 3287-3298.